PHY140Y

Spring Term – Tutorial 20 Discussion Solutions 6 March 2000

1. (a) We can solve this by direct substitution into the left-hand side of equation (1) in the discussion question. We find that

$$\frac{-\hbar^2}{2m} \left(\frac{\partial^2 \psi}{\partial x^2} + \frac{\partial^2 \psi}{\partial y^2} \right) + U(\vec{x})\psi = \frac{-\hbar^2}{2m} \left(\frac{\partial^2 \psi}{\partial x^2} + \frac{\partial^2 \psi}{\partial y^2} \right)$$
(1)

$$= \frac{-\hbar^2}{2m} \left(-k_x^2 - k_y^2\right)\psi \tag{2}$$

$$= \frac{\hbar^2}{2m} \left(k_x^2 + k_y^2\right) \psi \tag{3}$$

$$= E\psi, \qquad (4)$$

which means that ψ is a solution if we require that the total energy of the state be

$$E = \frac{\hbar^2}{2m} \left(k_x^2 + k_y^2 \right) \tag{5}$$

(b) We apply the same boundary conditions as before, namely that $\psi = 0$ at x = 0 and L and at y = 0 and L. The proposed solution already satisfies the boundary conditions at x = 0 and y = 0. The boundary conditions at x = L and y = L are satisfied if

$$\sin(k_x L) = 0 \tag{6}$$

$$\Rightarrow k_x L = n_x \pi \text{ where } n_x = 1, 2, 3, \dots$$
 (7)

$$\Rightarrow k_x = \frac{n_x \pi}{L}, \quad n_x = 1, 2, 3, \dots$$
(8)

$$\sin(k_y L) = 0 \tag{9}$$

$$\Rightarrow k_y L = n_y \pi \text{ where } n_y = 1, 2, 3, \dots$$
 (10)

$$\Rightarrow k_y = \frac{n_y \pi}{L}, \quad n_y = 1, 2, 3, \dots$$
(11)

(c) The three lowest lying energy states are

$$E_0 = \frac{\hbar^2 \pi^2}{2mL^2} (1+1) = \frac{\hbar^2 \pi^2}{mL^2}$$
(12)

$$E_1 = \frac{\hbar^2 \pi^2}{2mL^2} (1+4) = \frac{5\hbar^2 \pi^2}{2mL^2}$$
(13)

$$E_2 = \frac{\hbar^2 \pi^2}{2mL^2} \left(4+4\right) = \frac{4\hbar^2 \pi^2}{mL^2}$$
(14)

(15)

There is only one state with energy E_0 with $n_x = n_y = 1$. There are two states with energy E_1 , given by $n_x = 1$ and $n_y = 2$ and $n_x = 2$ and $n_y = 1$. There is only one state with energy E_2 : $n_x = n_y = 2$ (I leave the energy level diagram to the tutorial). Thus the degeneracies of these three states are 1, 2 and 1, respectively.



Figure 1: The probability distribution for the ground state of an electron in a 2-D box. I haven't bothered normalizing it (What is the normalization constant for this wave function?).

(d) In order for $E_0 = 10$ eV, we need

$$E_0 = \frac{\hbar^2 \pi^2}{mL^2} \tag{16}$$

$$\Rightarrow L = \sqrt{\frac{\hbar^2 \pi^2}{mE_0}} \tag{17}$$

$$= \sqrt{\frac{(6.58 \times 10^{-16})^2 \pi^2}{(5.11 \times 10^6)(10)}} = 2.89 \times 10^{-19} \text{ m}, \tag{18}$$

where I have expressed \hbar and the mass of the electron using units of electron Volts instead of Joules ($\hbar = 6.58 \times 10^{-16}$ eVs and $m = 5.11 \times 10^{6}$ eV).

- (e) The probability distribution for the ground state is just $|\psi(x, y)|^2$ and would look like the plot shown in Fig. 1.
- 2. (a) In the region $x \in (0, L)$, we have

$$\frac{-\hbar^2}{2m}\frac{d^2\psi_1}{dx^2} + U(x)\psi_1 = \frac{\hbar^2k_1^2}{2m}\psi_1$$
(19)

=

$$E\psi_1 \tag{20}$$

$$\Rightarrow E = \frac{\hbar^2 k_1^2}{2m} \tag{21}$$

(22)

and for the region x > L, we have

$$\frac{-\hbar^2}{2m}\frac{d^2\psi_2}{dx^2} + U(x)\psi_2 = \frac{-\hbar^2k_2^2}{2m}\psi_2 + U_0\psi_2$$
(23)

$$= E\psi_2 \tag{24}$$

$$\Rightarrow U_0 - E = \frac{\hbar^2 k_2^2}{2m}.$$
 (25)

Thus, ψ_1 and ψ_2 are solutions if the relationships between E, k_1 and k_2 are satisfied.

(b) We have already shown that

$$E = \frac{\hbar^2 k_1^2}{2m} \tag{26}$$

$$U_0 - E = \frac{\hbar^2 k_2^2}{2m}.$$
 (27)

(c) In order to tunnel as far as possible, we wish to make k_2 as small as possible (to reduce the variation in ψ_2 as a function of x). This can be achieved if we require that $U_0 - E$ be as small as possible.

The optimal value for L requires a little more thought. We have to consider the boundary conditions at x = L to determine this. In this case, the boundary condition of continuity of the wave function and its derivative requires that we do some work. These conditions imply that

$$A\sin(k_1L) = B\exp(k_2L) \quad \text{and} \quad Ak_1\cos(k_1L) = Bk_2\exp(k_2L) \tag{28}$$

$$\Rightarrow \sin(k_1 L) = \frac{k_1}{k_2} \cos(k_1 L) \tag{29}$$

$$\Rightarrow \tan(k_1 L) = \frac{k_1}{k_2}.$$
(30)

What this means is that for a fixed ratio of k_1 and k_2 (which is determined by the relative sizes of E and U_0), L is fixed by this continuity condition, modulo a factor of $n\pi/k_1$ where n can be a positive integer. Since we are trying to find solutions that have ψ_2 as large as possible, the optimal choice for L is one that minimizes k_2L (the argument to the exponential in ψ_2), or the smallest value of L that also satisfies the condition in Eq. 30.

(d) We have already looked at the impact of the boundary conditions on k_1 and k_2 . For given L and U_0 , the condition in Eq. 30 determines the allowed energy levels. Once we have those, we then have to require that there be continuity at x = l, namely

$$A\sin(k_1L) = B\exp(k_2L) \tag{31}$$

$$\Rightarrow B = A \frac{\sin(k_1 L)}{\exp(k_2 L)} \tag{32}$$

and that the entire wave function is normalized:

$$1 = \int_0^L |\psi_1^2| dx + \int_L^\infty |\psi_2^2| dx$$
 (33)

$$= \frac{A^2}{k_1} \int_0^{k_1 L} \sin^2 z dz + \frac{B^2}{k_2} \int_{2k_2 L}^\infty \exp(w) dw$$
(34)

$$= A^{2} \left[\frac{1}{k_{1}} \int_{0}^{k_{1}L} \sin^{2} z dz + \frac{\sin^{2}(k_{1}L)}{k_{2} \exp(2k_{2}L)} \int_{2k_{2}L}^{\infty} \exp(w) dw \right]$$
(35)

$$\Rightarrow A = \left[\frac{1}{k_1} \int_0^{k_1 L} \sin^2 z dz + \frac{\sin^2(k_1 L)}{k_2 \exp(2k_2 L)} \int_{2k_2 L}^\infty \exp(w) dw\right]^{-1/2}.$$
 (36)

3. The uncertainty in the electron's vertical position, Δy , leads to an uncertainty in the vertical component of their momentum, Δp_y , that must satisfy

$$\Delta p_y \ge \frac{\hbar}{\Delta y}.\tag{37}$$

Since the electrons have a horizontal momentum of p_x , we can determine the minimum spread in the electron beam, $\Delta \theta$, by noting that

$$\tan \Delta \theta = \frac{\Delta p_y}{p_x} \tag{38}$$

$$\geq \frac{\hbar}{p_x \Delta y} \tag{39}$$

$$= \frac{\hbar}{m_e v_x \Delta y} = \frac{1.055 \times 10^{-34}}{(9.11 \times 10^{-31})(2.2 \times 10^7)(9 \times 10^{-8})}$$
(40)

$$= 5.85 \times 10^{-5}$$
 radians. (41)

Note that the Heisenberg Uncertainty Principle applies to each coordinate separately!