

# SYSTEMATICS OF NUCLEAR PROPERTIES

7

ATOMIC NUMBER  $Z$  CORRESPONDS TO POSITION IN PERIODIC TABLE, HENCE CHARGE ON THE NUCLEUS

ATOMIC MASS NUMBER  $A$  NEAREST INTEGER TO THE RATIO OF MASS OF ATOM, TO MASS OF ATOM OF LIGHTEST ISOTOPE OF H

ATOMIC MASS RATIO OF MASS OF ATOM, TO  $\frac{1}{12}$  OF MASS OF ATOM OF CARBON  $A=12$

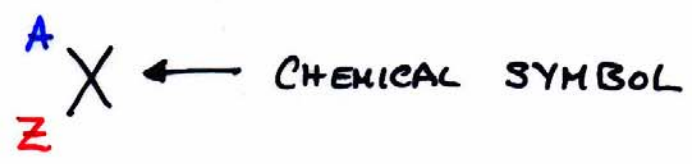
OR  $\frac{1}{16}$  OF MASS OF ATOM OF OXYGEN  $A=16$

(SLIGHTLY MORE COMPLEX)  $\rightarrow$

ATOMIC WEIGHT ATOMIC MASS AVERAGED OVER TERRESTRIAL ISOTOPE ABUNDANCE FOR ELEMENTS WITH MORE THAN ONE STABLE ISOTOPE

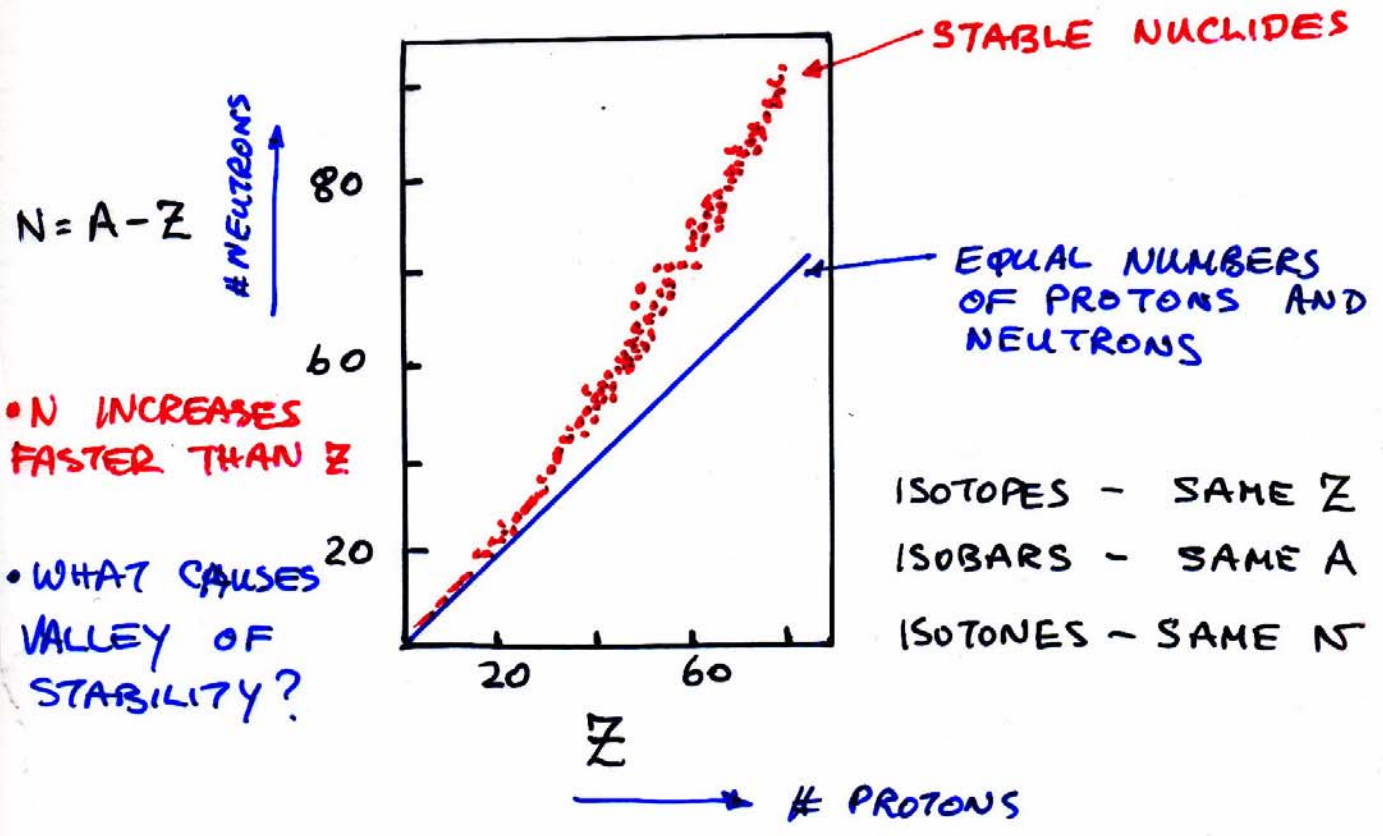
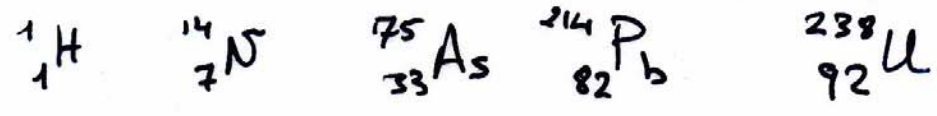
ASSUMING THAT A NUCLEUS CONTAINS  
PROTONS & NEUTRONS

A GIVEN SPECIES (NUCLIDE) IS CHARACTERIZED!



$Z = \text{ATOMIC NUMBER} = \text{NO. OF PROTONS} = \#e^-$   
 $N = \text{NUMBER OF NEUTRONS}$  ↑  
ATOM

$A = \text{MASS NUMBER} = Z + N = \#p + \#n$



IF ONE KNEW NOTHING ABOUT SPECIAL RELATIVITY  
- ONE MIGHT EXPECT

$$M(A, Z) = Z m_p + (A - Z) m_n$$

MASS OF NUCLEUS      PROTON MASS  $938.27 \frac{\text{MeV}}{c^2}$       NEUTRON MASS  $939.56 \frac{\text{MeV}}{c^2}$

- NUCLEUS IS STABLE, HAVE TO DO WORK TO PULL IT APART → BOUND STATE

$$M(A, Z) < Z m_p + (A - Z) m_n$$

$$\Delta M(A, Z) = M(A, Z) - \underbrace{Z m_p + (A - Z) m_n}_{\text{MASS OF CONSTITUENTS AT INFINITY}}$$

MASS DEFICIT      MASS OF BOUND STATE

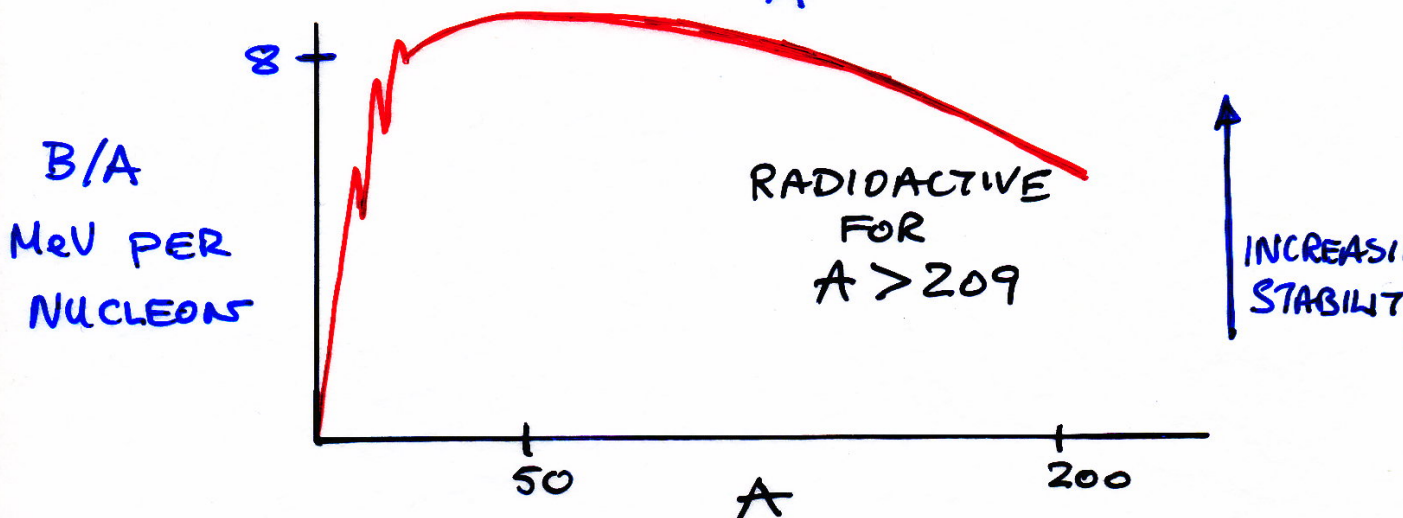
$$-\frac{BE}{A} = -\frac{\Delta M(A, Z) c^2}{A} = \text{BINDING ENERGY PER NUCLEON}$$

BINDING ENERGY IS ENERGY ONE HAS TO PUT INTO SYSTEM TO REMOVE CONSTITUENTS TO INFINITY.

TOTAL BINDING ENERGY WILL DEPEND ON TOTAL NUMBER OF NUCLEONS

$\frac{BE}{A} \Rightarrow$  EFFECT ON INDIVIDUAL NUCLEON

$$\frac{B}{A} = \frac{(Z m_p + (A-Z) m_n - M(A, Z)) c^2}{A}$$



- INCREASES RAPIDLY  $\rightarrow$  SPIKES  $\leftarrow$  VERY STABLE
  - PEAKS AT 9 MeV/NUCLEON
  - DROPS SLOWLY  $\leftarrow$  ROUGHLY CONSTANT AT 8 MeV/NUCLEON
- SATURATION IS AN IMPORTANT INDICATOR OF BEHAVIOUR OF FORCE BETWEEN TWO NUCLEONS.
- FISSION / FUSION  $\Rightarrow$  ENERGY RELEASE

## MARGINAL NOTE ON MASSES

- NUCLEAR MASS  $\approx$  ATOMIC MASS -  $Zme$

NEGLECT ELECTRON BINDING ENERGY

- 16 amu = MASS OF  ${}^{16}_8\text{O}$   $\leftarrow$  OLD SCALE
- 12 u = MASS OF  ${}^{12}_6\text{C}$   $\leftarrow$  NEW SI MASS SCALE
- ONE amu = MASS IN GRAMS OF ONE "ATOM OF ELEMENT WITH ATOMIC WEIGHT = 1.0000"

AVOGADRO = # OF ATOMS IN MOLE  
MOLE = ATOMIC WT IN GRAMS  
 $A_0$  NUCLEI IN A MOLE  
MOLE HAS MASS A

$$\therefore \text{MASS OF NUCLEUS} = \frac{A}{A_0}$$

- MASS OF FICTIONAL ATOM WITH ATOMIC WEIGHT OF 1.0000 gm =  $A_0^{-1}$

AVOGADRO SETS SCALE OF amu.

$m_p = 1.00728 \text{ amu}$

$1 \text{ amu} = \frac{938.27 \text{ MeV}}{1.00728 c^2}$

$1 \text{ amu} = 1.6726 \times 10^{-24} \text{ gm}$

THIS IS ON THE OLD <sup>16</sup>O BASED SCALE  
ON THE SI - "UNIFIED CARBON SCALE"

$1u = 931.49432 \cdot \text{MeV}/c^2$   
 $= 1.660540 \times 10^{-27} \text{ kg}$

$m_p = 938.27 \frac{\text{MeV}}{c^2} \cdot \frac{1}{931.494 \text{ MeV}} \cdot c^2 \cdot u$

$m_p = 1.00727 u$

$m_p$  IS THE SAME PHYSICAL VALUE !  
JUST SLIGHTLY DIFFERENT STANDARD  
FOR UNITS

# DENSITY OF NUCLEAR MATTER

MEASUREMENTS OF NUCLEAR SIZES IN SCATTERING EXPERIMENTS SHOW

$$R = R_0 A^{\frac{1}{3}} \sim 1.2 \times 10^{-13} A^{\frac{1}{3}} \text{ cm}$$

$\uparrow$  CONSTANT = 1.2 A<sup>1/3</sup> fm

SINCE VOLUME =  $\frac{4}{3} \pi R^3 = \frac{4}{3} \pi R_0^3 A$

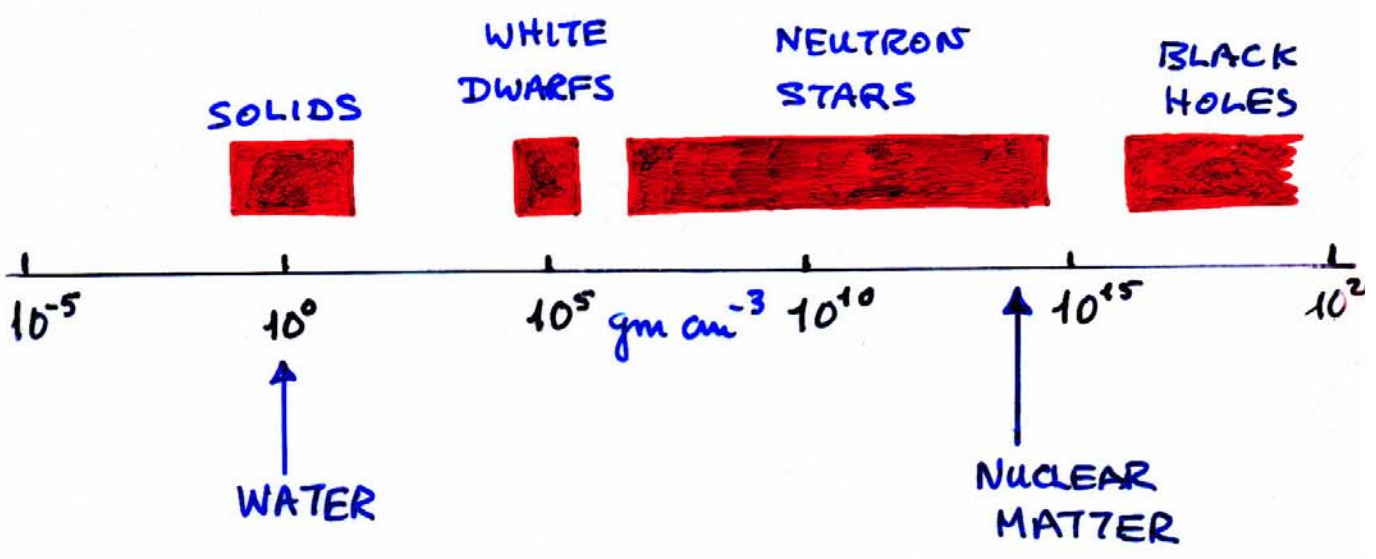
VOLUME  $\propto$  NUMBER OF NUCLEONS

→ NUCLEONS TIGHTLY PACKED IN NUCLEUS

$$\rho = \frac{\text{MASS}}{\text{VOLUME}} = \frac{A \cdot 1.660 \times 10^{-24} \text{ gm}}{\frac{4}{3} \cdot \pi \cdot R_0^3 \cdot A}$$

$$\approx \underline{2 \times 10^{14} \text{ gm. cm}^{-3}}$$

} FOR ALL A



# NUCLEAR INSTABILITY

AWAY FROM THE VALLEY OF STABILITY  
NUCLEI ARE UNSTABLE

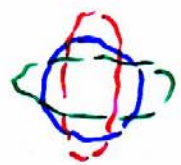


$\alpha$  : BOUND SYSTEM OF  ${}^4_2\text{He}$  "tunnels"  
OUT OF NUCLEUS

$\beta$  : NUCLEUS EMITS AN ELECTRON

$$n \rightarrow p + e^- + \bar{\nu}_e$$

$\gamma$  : NUCLEUS EMITS A PHOTON



CHARGED BODY OF OSCILLATING  
SHAPE CAN EMIT  
ELECTROMAGNETIC  
RADIATION



# NUCLEAR MAGNETIC MOMENT

SPINNING  
CHARGE

ELECTRIC

(4)

MAGNETIC DIPOLE  
MOMENT

$$\vec{\mu} = g \cdot \frac{e}{2mc} \cdot \vec{S}$$

LANDÉ  
g-FACTOR

MASS

SPIN

BOHR  
MAGNETON

DIRAC  
FOR A POINT PARTICLE

$$g = 2$$

$$\mu_e = \mu_B$$

FOR A PARTICLE WITH  
SUBSTRUCTURE

$$g \neq 2$$

$$\mu_p = \frac{\mu_B}{2000}$$

$$\mu_B = e\hbar/2mc$$

$$\mu_N = e\hbar/2mpc$$

$$\mu_p = 2.79\mu_N; \mu_n = -1.91\mu_N$$

OBSERVE MAGNETIC MOMENTS OF  
NUCLEI  $\sim \mu_N \approx 0.3\mu_p$

- IF ELECTRONS IN NUCLEI

NITROGEN

$$\mu \sim 14\mu_p + 7\mu_e$$

THOUSANDS OF TIMES LARGER  
THAN OBSERVED

# NATURE OF THE NUCLEAR FORCE

- CAVEAT - BY NUCLEAR FORCE, MEAN FORCE BETWEEN NUCLEONS
- WILL LEARN LATER THAT IT IS A RESIDUAL EFFECT OF THE FORCE BETWEEN QUARKS

FORCE THAT BINDS NUCLEONS INTO NUCLEI

- CANNOT BE A KNOWN CLASSICAL FORCE
  - GRAVITY TOO WEAK BY MANY ORDERS OF MAGNITUDE
  - EM FORCE COULD ONLY BIND PROTONS & NEUTRONS THROUGH INTERACTION OF MAGNETIC MOMENTS → AGAIN TOO WEAK
  - COULOMB FORCE TENDS TO DISRUPT NUCLEUS

# SHORT RANGE FORCE

- FACT THAT ATOMIC STRUCTURE CAN BE VERY ACCURATELY CALCULATED ASSUMING ONLY THE COULOMB FORCE
  - NUCLEAR FORCE DOES NOT EXTEND BEYOND NUCLEI

- BINDING ENERGY PER NUCLEON IS ROUGHLY CONSTANT AS A INCREASES
  - BINDING ENERGY DOES NOT DEPEND UPON SIZE OF NUCLEUS
  - IF NUCLEAR FORCE WAS LONG RANGE LIKE COULOMB

$\frac{1}{2} (A-1)A$  PAIRWISE COMBINATIONS

BINDING ENERGY ~ TOTAL POTENTIAL ENERGY OF ALL COMBINATIONS

$B \propto A(A-1)$

FOR  $A \gg 1 \rightarrow B \propto A^2$

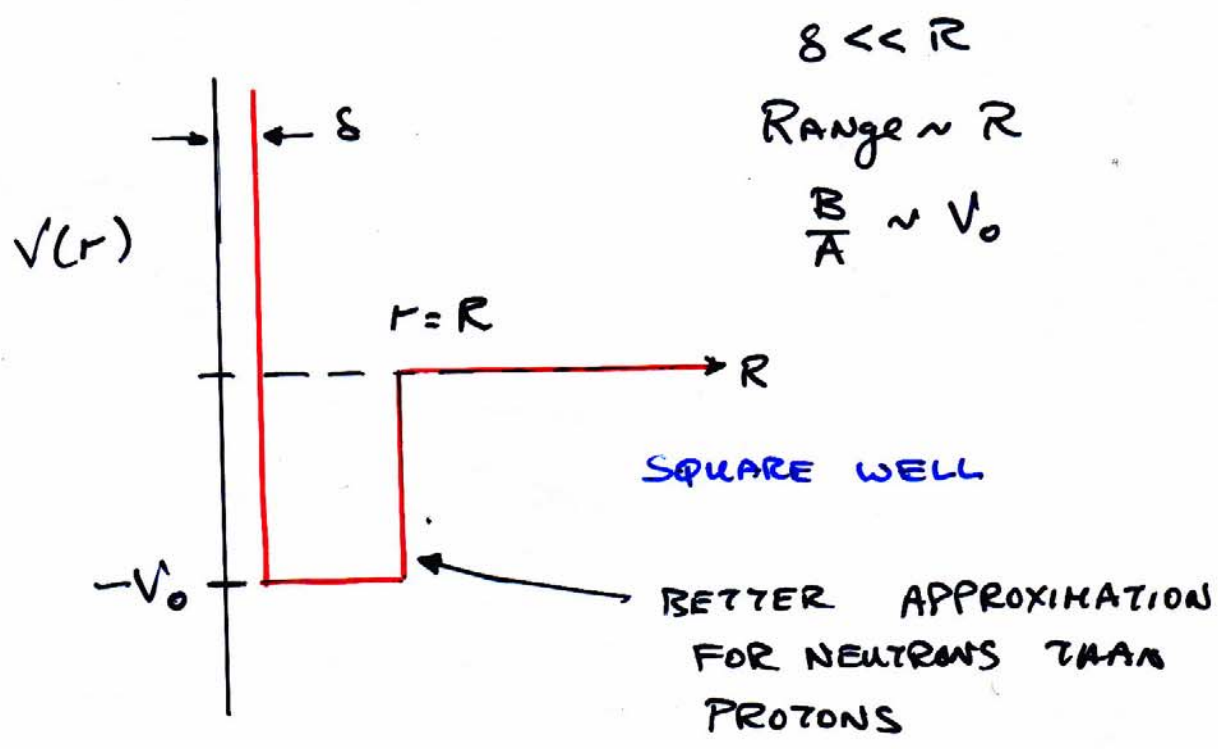
BINDING ENERGY PER NUCLEON =  $\frac{B}{A} \propto A$  ← LONG RANGE FORCE

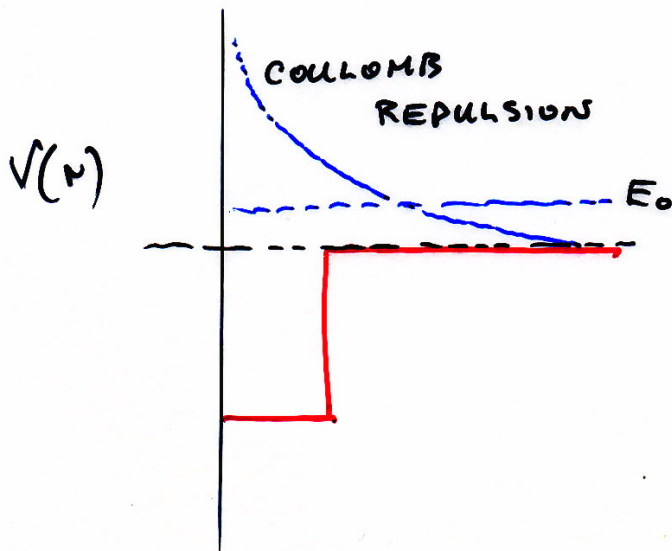
LONG RANGE FORCE  $\frac{B}{A} \propto A$

EXPERIMENTALLY  $\frac{B}{A}$  ROUGHLY CONSTANT

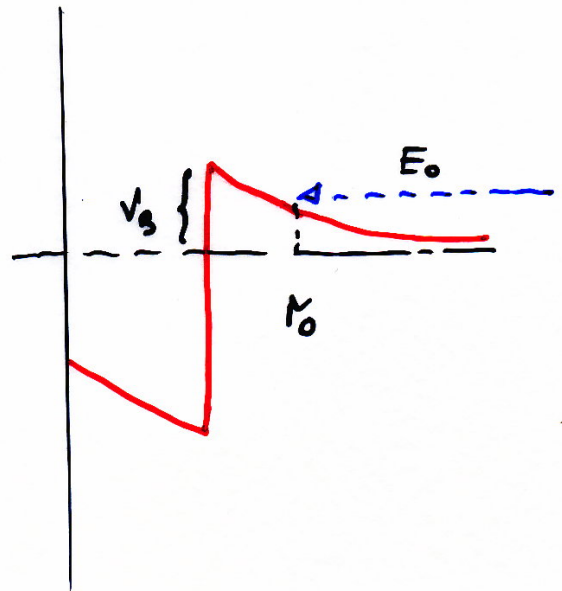
→ FORCE SATURATES

- A GIVEN NUCLEON CAN ONLY INTERACT WITH ITS NEAREST NEIGHBOURS
- ADDING MORE NUCLEONS INCREASES SIZE OF NUCLEUS - BUT NOT BINDING ENERGY/NUCLEON
- ATTRACTIVE FORCE, BUT MUST HAVE A REPULSIVE CORE IN ORDER TO PREVENT COLLAPSE - INTERNAL STRUCTURE OF NUCLEONS





POTENTIAL  
SEEN BY NEUTRON

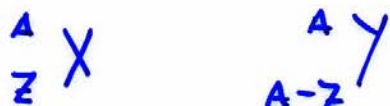


POTENTIAL  
SEEN BY  
PROTON

- A NEUTRON JUST SNEAKS INTO THE NUCLEUS, AND FALLS INTO THE POTENTIAL WELL.
- A PROTON SEES A REPULSIVE FORCE. MINIMUM DISTANCE OF PENETRATION
- SCATTERING EXPERIMENTS INSENSITIVE TO EXACT FORM OF NUCLEAR POTENTIAL
- FROM ELEMENTARY QUANTUM MECHANICS WE EXPECT THE SQUARE WELL TO LEAD TO DISCRETE ENERGY LEVELS  
⇒ LIKE ATOMS

# NON CLASSICAL FEATURES OF NUCLEAR FORCE

FIND "MIRROR NUCLEI", WHERE



- HAVE SAME NUMBER OF  $n-p$  INTERACTIONS BUT DIFFERENT NUMBER OF  $pp$   $nn$
- AFTER CORRECTING FOR COULOMB EFFECTS  
FORCE BETWEEN  $nn$  = FORCE BETWEEN  $pp$   
→ CHARGE INDEPENDENCE OF NUCLEAR FOR
- NEUTRONS & PROTONS ARE TWO "CHARGE" STATES OF ONE PARTICLE → NUCLEONS
- NUCLEON CAN HAVE TWO ORIENTATIONS IN "CHARGE SPACE"

THIS LEADS TO CONCEPT OF ISOTOPIC SPIN SYMMETRY

→ NUCLEAR INTERACTIONS INVARIANT UNDER "ROTATIONS IN ISOSPIN SPACE"

# PAIRING OF NUCLEONS

EASY TO SEE (SINCE  $\vec{n}, \vec{p} = \frac{\hbar}{2}$ )

EVEN A → INTEGRAL NUCLEAR SPIN

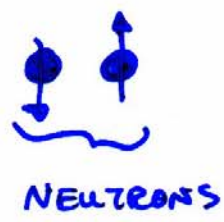
ODD A →  $\frac{1}{2}$  INTEGRAL NUCLEAR SPIN

- STRIKING EXPERIMENTAL FACT IS THAT ALL ~~NUCLEONS~~ NUCLEI

EVEN PROTONS } ZERO NUCLEAR SPIN  
 EVEN NEUTRONS }  
 EVEN - EVEN

- ALSO LARGE NUCLEI (MANY  $\frac{\hbar}{2}$ ) GENERALLY HAVE VERY SMALL NUCLEAR SPINS

QUANTUM MECHANICAL PHENOMENON  
 → NUCLEONS PAIR UP



$\uparrow \frac{\hbar}{2}$  TWO LIKE NUCLEONS BIND MORE TIGHTLY THAN TWO UNLIKE NUCLEONS

- THIS IS CONSISTENT WITH FACT THAT NUCLEAR MAGNETIC MOMENTS ARE SMALL

$$-3\mu_N \rightarrow +10\mu_N$$